

## Identification Of Chemistry Bioactivity Structure Of $\alpha$ -Glucosidase Inhibitor From Ethanol Extract Of The Stem Bark Raru (*Vatica Pauciflora* Blume)

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**Abstract:** Inhibitory compound of  $\alpha$ -glucosidase bioactivity was isolated from Ethanol extract of the stem bark of Raru (*Vatica pauciflora* Blume) which, is a wild plant that grows in Tapanuli state in Indonesia. The ethanol extract of the stem bark of raru was partitioned and was chromatographed on columns with stationary phase silica gel 60 mesh  $F_{254}$  (0,063 mm – 0,200 mm). The ethanol extract of isolates showed the enzyme  $\alpha$ -glucosidase inhibitor  $IC_{50}$  of 93.46. The next step is the determination of the chemical structure of the UV Vis, FT-IR, NMR, (COSY, HMQC and HMBC). It found of compounds consist of 2 methoxy, 1 aromatic, and 1 carbonil compounds. The Isolat is a 3,4,9, Trihidroxy-2-(hydroxymethyl)-8-10-dimethoxy-2-3-4 tetrahydropyrano(3,2-c)isochromen-6(10bH)-one according to IUPAC.

**Keywords:** Raru (*Vaticapauciflora* Blume), antidiabetic,  $\alpha$ -glucosidase inhibitor

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### I. Introduction

The use of traditional medicine is Indonesian culture that goes back thousands of years ago. It is a cultural heritage of Indonesia that need to be explored, researched, and developed (Hedi, 2007). Consumed plant extracts for the treatment, which is made in the form of decoction, herbal remedies and herbal capsule form. Things like this do to treat diabetes mellitus.

Diabetes mellitus (DM) is a condition of the concentration of glucose in the blood is chronically higher than normal value (hyperglycemia). This is caused by a deficiency of insulin or the insulin does not function effectively. Diabetes can lead to various diseases such as hypertension, stroke, coronary heart disease, and kidney failure (Guyton, and Hall, 2007).

Measurement of glucose levels can be determined in vitro by enzymatic methods (Lucile, 1997). Spectrophotometer for color intensity is used hereinafter blood glucose levels can be determined.

According to WHO (2005), more than 80% of the population in developing countries consume natural ingredients derived mainly from plants, either as a form of maintaining health and medicine. The use of natural medicine is intended to avoid the side effects of chemical drugs. And plant-based treatments currently has a market share of about 30%.

Adebayo (2008) found extracts of plants as antidiabetic impact activities such as: *Vermoniaamygdalina*, *Bidenspilosa*, *Carica papaya*, *Citrus aurantiifolia*, *Ocimumgratisimum*, *Momordica charantia* and *Morindalucida* these plants have been consumed in Nigeria.

Likewise Gunawan (2009) discovered a flavonoid in 4 raru plant species, namely (1) *Cotylelobiummelanoxyllum* Pierre, (2) *Shoreabolancarpoides* Symington, (3) *Cotylelobiumlanceolatum* craib, and (4) Pierre *Cotylelobiummelanoxyllum* be expressed that these compounds can lower blood sugar levels in vitro.

There are three compounds that have activity flafonoid as antidiabetic, namely (1) 3- $\beta$ -hidroksynaringenin or isoaromadendrin, (2) taxifolin, and (3) 5-hidroxy-3, 4', 7 trimethoksiflavanone. Flavonoids extracted with ethanol. Compounds 1 and 2 were found from plant *Euphorbia cuneata* Vahl, and compounds to the three found in the ethanol extract of *Salvia L. verbenaca* Flavonoids were identified by spectroscopic data  $^1H$ ,  $^{13}C$  NMR, DEPT, COSY, HMQC and NOESY (Bahar, 2005). Isolation of ethyl acetate extract from the leaves of plants *Arto carpus communis* flavonoid found to have activity as an antidiabetic, ie 8-geranyl-4,5,7-flavone trihidroksi. This can inhibit  $\alpha$ -glucosidase enzyme  $IC_{50}$  18.12  $\mu$ g mL<sup>-1</sup> (Puspa, 2008).

In the ethanol extract of the stem bark raru contained flavonoids are phytochemicals. Antidiabetic bioactivity barrier test conducted by Ida Riris Duma (2013) in vitro against hexane, ethyl acetate, ethanol, and water from the plant stem bark raru (*Vatica pauciflora* Blume), obtained bioactivity of  $\alpha$ -glucosidase inhibition of ethanol extract of the stem bark extract is higher than ethyl acetate, hexane and water by using acarbose as a control. In this study bark raru types *Vatica pauciflora* Blume extracted, and further testing of the inhibitory activity against enzim  $\alpha$ -glucosidase. Toxicity tests conducted by Brine Shrimp Lethality Test (BSLT). Further isolation and elucidation performed on extracts that have inhibitory activity of  $\alpha$ -glucosidase enzyme of the most high. The chemical structure determined by spectroscopic spectral data (UV, FT IR1, NMR1D, 2D NMR (COSY, HMQC, HMBC), and HR MS.

## II. Methods And Materials

Raru bark samples (*Vaticapauciflora* Blume) extracted by solvent extraction method stratified by different levels of polarity, namely n-hexane, ethyl acetate, 96% ethanol, and water. Each extract was concentrated, tested toxicity with BSLT method. With this method, the toxicity of a compound with LC50 value is the concentration of the compound that gives as much as 50% mortality rate. Active compounds will provide a high mortality. The smaller the value, the greater the toxicity LC50. A sample said to be highly toxic to shrimp larvae when *Artemiasalina* Leach has LC50 <30 mg / mL. LC50 has declared toxic when 30-1000 mg / mL and less toxic when having LC50 > 1000 mg / mL (Steven and Russell, 1993).

Bioactivity inhibition of the enzyme  $\alpha$ -glucosidase carried out in vitro by the method of Kawanishi, that inhibition of solving the model substrate p-nitrophenol- $\alpha$ -D-glukopiranosid to p-nitrophenol and glucose yellow. The enzyme activity was measured based on the absorbance of p-nitrophenol right generated by UV-visible spectrophotometry at  $\lambda$  400 nm (Matsui, 2001). Phytochemical test was conducted using Harborn 1987. Extract that has the highest bioactivity separated by column chromatography. The shape of the stem bark raru extraction scheme shown in Figure 1.

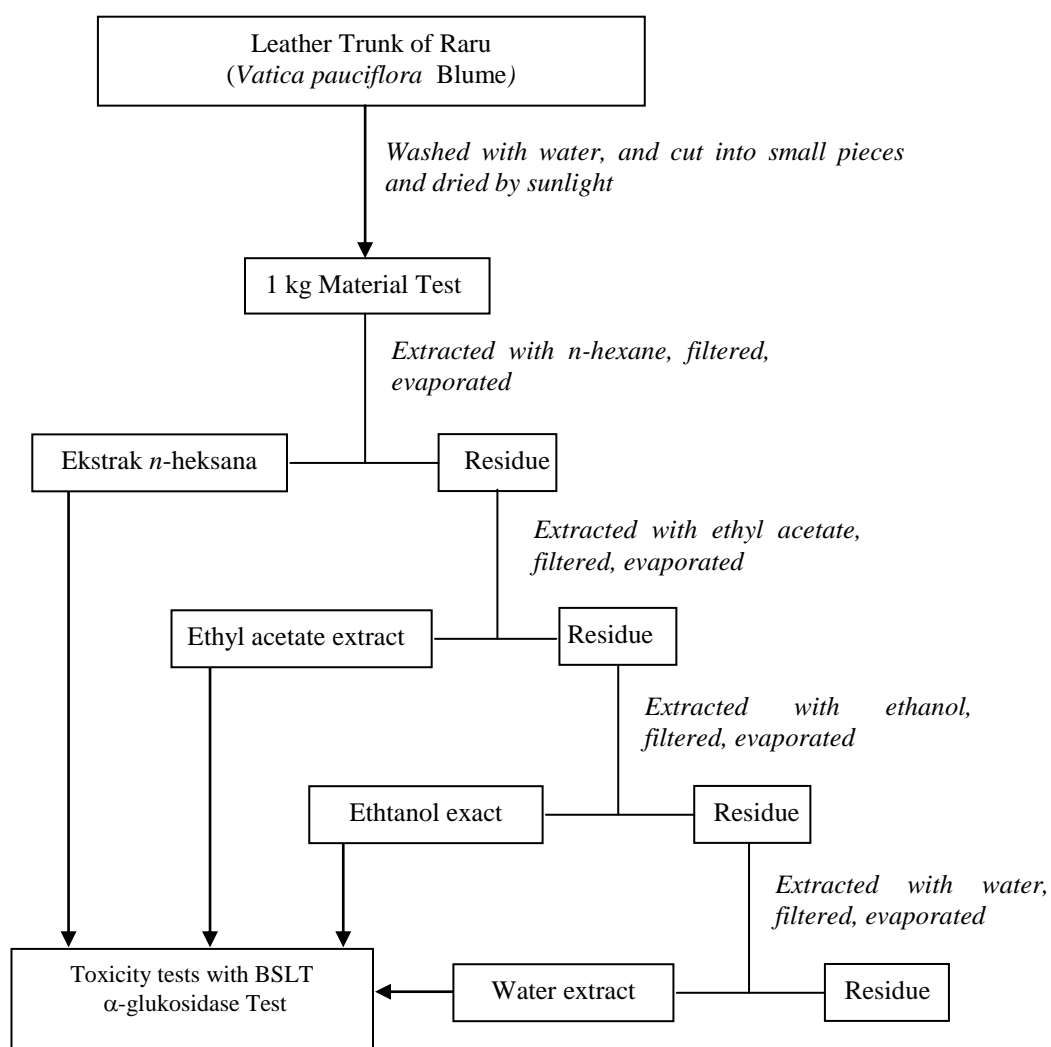


Figure 1. Schematic extraction, fractionation of the stem bark of raru (*Vatica pauciflora*)

### Fractionation By Column Chromatography Method

Extract Ethanol Fractionation by Chromatography column I, performed by taking as much as 5.0 g of ethanol extract, and further fractionated by column chromatography using silica gel stationary phase 60 mesh. And the mobile phase used was chloroform-methanol (40:1 ~ 1:1). The compounds in botanicals, grouped by polarity in the form of fractions. To obtain a more simple fractions, done by combining fractions that have the same pattern of the chromatogram. Results fractions of column 1 are combined into 14 fractions, and then tested

the inhibition of the enzyme  $\alpha$ -glucosidase. The test results showed inhibition of the enzyme  $\alpha$ -glucosidase is the fraction 9.

Extract Ethanol Fractionation by Column Chromatography II made to the fraction 9 that the same method as in column I. Results fractions were combined, and obtained six fractions, were then tested for inhibition against  $\alpha$ -glucosidase enzyme.

Further separation of fraction VPET-9 with Chromatography Columns III conducted in fractions 9-4 with the same method as in column I. Results fractions were combined and gained 5 fractions. Tests conducted on the inhibition of the enzyme  $\alpha$ -glucosidase. Provided that fractions 9-4 VPET-4 of the ethanol extract had  $\alpha$ -glucosidase inhibitory activity of the most high, and further separated by column chromatography. This was done to obtain a pure compound.

Purity test using 2-dimensional TLC performed on fractions VPET-9-4-4 obtained from test results antidiabetic. Analysis was performed using silica gel stationary phase GF<sub>254</sub> plates with a mobile phase of chloroform-methanol (2:1), and chloroform-acetonitrile (2:1). Chemical structure determination carried out by UV-Vis spectroscopy, FT-IR, NMR analysis (<sup>1</sup>H-<sup>13</sup>C-NMR and DEPT) and 2D NMR (COSY, HMQC, HMBC), and the number of isolates VPET-9-4-4 reconstituted with CD<sub>3</sub>OD .

### III. Result

The yield of extraction results shown in Table 1, where the highest obtained in the extraction of ethanol.

Table 1. The results of the yield of the extract of the stem bark of raru (*Vaticapauciflora* Blume)

Sample	Weight (g)	The yield of (%) *
<i>n</i> -heksana	6,21	0,62
Etilasetat	58,62	5,86
Ethanol	76,13	7,61
Distilled water	19,47	1,95

Description: \* 1 kg calculated on the dry crude drug

The result of extract toxicity test of *n*-Heksana, Etil Asetat, Etanol and water dan Water by BSLT method, each of them is *n*-heksan extract LD<sub>50</sub> = 368,51 ppm, etil acetate = 19,45, ethanol = 5,76 ppm, dan air = 36,22 ppm. It shown that ethanol extract has the least toxicity. The results of Antidiabetic test of each extract by  $\alpha$ -glucosidase inhibitory method is shown in table 2.

Table 2. Test results antidiabetic extract using an enzyme inhibition method  $\alpha$ Glukosidase

Ekstrak	Inhibisi (%)
<i>n</i> -heksana	28,98
Etilasetat	60,83
Ethanol	91,08
Water	78,34

Phytochemical test results on ethanol extract of bark which has bioactivity antidiabet raru highest, with Harborn method, indicates that there are flavonoids, saponins, quinones, tannins, triterpenoids, minyakatsiri, and coumarin.

The results of the analysis of NMR (Nuclear Magnetic Resonance) 1-dimensional (<sup>1</sup>H and <sup>13</sup>C-NMR) is based on Silverstein (1991). Proton spectra (<sup>1</sup>H-NMR and carbon (<sup>13</sup>C-NMR) was done by dissolving the samples in deutereum methanol (CD<sub>3</sub>OD) (0.5 mL), respectively in the NMR tube (5 mm). Spectra were recorded on a JEOL 500 spectrophotometer (<sup>1</sup>H-NMR MHz and <sup>13</sup>C-NMR at 125 MHz).

<sup>1</sup>H-NMR spectra of ethanol extracts To Isolate Isolate 9-4-4-1. Proton Nuclear Magnetic Resonance spectra (<sup>1</sup>H-NMR) indicate the type and number of protons contained in the ethanol extract of isolates 9-4-4-1. Proton methoxy (OCH<sub>3</sub>) in the area of chemical shift (chemical shift)  $\delta$ H 3.87 and 3.97 (3H) in the singlet splitting pattern (s). Some metin proton (CH) are the chemical shifts  $\delta$ H 3.53; 3.54; 3.77; 3.82; 3.96; 3.99, and 4.80, which is characteristic for the protons that resonate with the oxygen atom, and one olefinic proton at  $\delta$ H metin that 7,32 (s), which does not resonate with other protons.

<sup>13</sup>C-NMR spectra To Isolate 9-4-4-1. <sup>13</sup>C-NMR spectra and DEPT experiments analysis (Distortionless Enhancement by Polarization Transfer) shows there are 15 carbon on the chemical structure of the ethanol extract had inhibition against  $\alpha$ -glucosidase enzyme highest as shown in Figure 2.

The carbonyl atom located very down field that is at  $\delta$ C 166.15 (s). Methoxy peaks are at  $\delta$ C 61.43 (q) and 62.02 (q). The presence of the chemical shift in the high field region is about  $\delta$ C 62.6 (t) and 71.71 (d), 73.26 (d), 76.05 (d), 81.91 (d), 82.70 (d), all of which resonate carbon atoms with oxygen atoms. And for the aromatic carbon present in  $\delta$ C 115.00 (d), 120.62 (s), 125.97 (s), 148.72 (s), 152.64 (s) and 152.72 (s). The

relationship between carbon and proton in the ethanol extract of the isolates showed no association between the signal and the carbon nucleus on the same proton. Spot spectrum indicates the two are directly related to the bond. Proton signal at  $\delta$ H 3.79 ppm seen (H1-13) and  $\delta$ H 3.99 (H2-13). C-12 at  $\delta$ C 73.26 with  $\delta$ H 4.80 (H-12), C-3 at  $\delta$ C 115.00 with  $\delta$ H 7.32 (H-3).

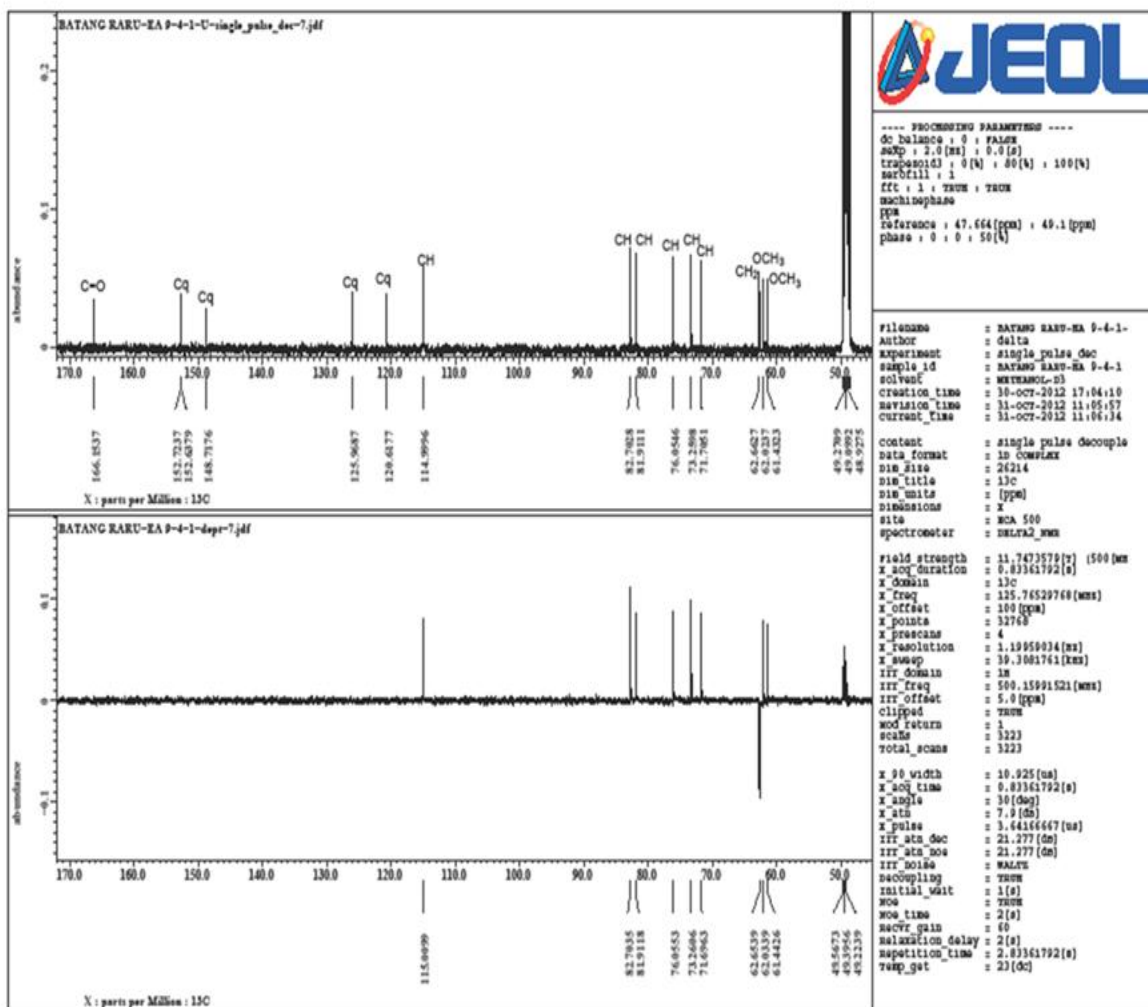


Figure 2. DEPT 13C-NMR spectrum of isolates

The relationship between protons and protons in the chemical structure of 9-4-4-1 isolates appear from analysis of COSY spectra. The correlation suggests that the protons of methylene (CH<sub>2</sub>)  $\delta$ H 3.79 (H1-13) and 3.99 (H2-13) associated with the proton at  $\delta$ H 4.80 (H-12);  $\delta$ H 4.80 (H-12) with  $\delta$ H 3.96 (H-11). Likewise  $\delta$ H 3.53 (H-10) with  $\delta$ H 3.82 (H-9);  $\delta$ H 3.82 (H-9) with  $\delta$ H 3.54 (H-8).

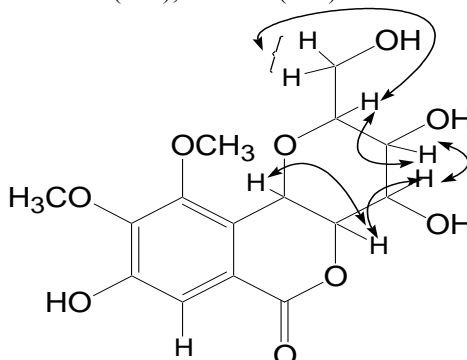


Figure 3. COSY spectra analysis results for isolates Chemical Structure 9-4-4-1

The results of the analysis of HMBC spectra of carbon visible at the position C-1 ( $\delta_C$  166.15) C-2 ( $\delta_C$  125.97), C-4 ( $\delta_C$  152.64), C-5 ( $\delta_C$  152.72) and C-7 ( $\delta_C$  120.62). The existence of long-range HMBC Correlation of the spectra looks like Figure 3.

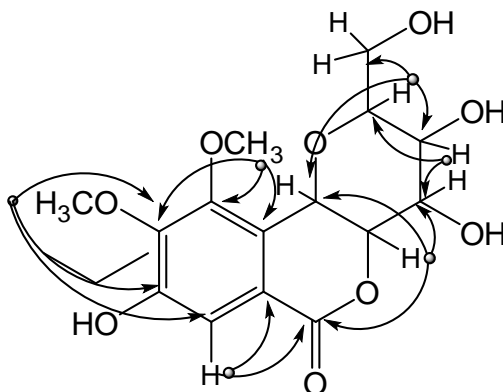


Figure 4. Isolates chemical structure 9-4-4-1 HMBC analysis results

Chemical shifts for proton and carbon chemical structure isolates 9-4-4-1 shown in Figure 4

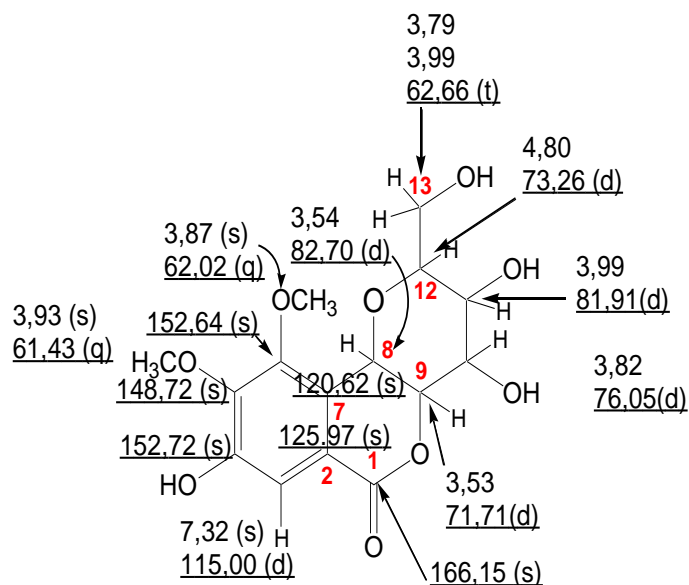


Figure 5. Proton and Carbon Chemical Shifts for Chemical Structure Isolates 9-4-4-1. Carbon chemical shift data for the proton and the chemical structure of 9-4-4-1 isolates presented in Table 3 below.

Table 3. Carbon Chemical Shift Correlation and H-13 C NMR To Isolate 9-4-4-1 Based on 2D NMR HMQC

No.	$\delta_C$ (ppm)/DEPT	$\delta_H$ (ppm)	No Urut Karbon Sesuai Struktur Kimia
1.	61,43 (q)	3,97	Ome
2.	62,02(q)	3,87	Ome
3.	62,66(t)	3,79;3,99	C-13
4.	71,71(d)	3,57	C-9
5.	73,26(26)	4,80	C-12
6.	76,05(d)	3,82	C-10
7.	81,91(d)	3,99	C-11
8.	82,70(d)	3,54	C-8
9.	115,00(d)	7,32	C-3
10.	120,62(s)	-	C-7
11.	125,97(s)	-	C-2
12.	148,72(s)	-	C-5
13.	152,64(s)	-	C-6
14.	152,72(s)	-	C-4
15.	166,15(s)	-	C-1

Analysis of mass spectroscopy (MS) is conducted by the HR-MS instrument (High Resolution Mass Spectroscopy), showed a molecular ion at  $m/z$  342 (M)<sup>+</sup>. This indicates isolates 9-4-4-1 has a molecular weight (MW = 342) for the molecular formula mass C<sub>15</sub>H<sub>18</sub>O<sub>9</sub>. Spektra isolate compounds 9-4-4-1 gave molecular ion peaks and fragment ions in the spectra of  $m/z$ : 325 ; 265; 165 and 121.

#### IV. Conclusion

Isolation and structure elucidation of chemical that has activity as an inhibitor of  $\alpha$ -glucosidase enzyme from the stem bark raru (*Vatica pauciflora* Blume) concluded that, (1) Bark raru efficacious as antidiabetic. Inhibit the activity of the enzyme  $\alpha$ -vitro glukosidasesecara VpEt fraction 9-4-4-1, that is equal to IC<sub>50</sub> 93.46. (2) The chemical structure of spectral data based on UV-Vis spectrophotometer, FT-IR and NMR spectrometer dankarbon proton NMR 1D) and 2D (COSY, HMQC, and HMBC); HRMS isolates obtained VpEt 9-4-4-1 is bergenin dimethoxy compound. The Isolate is a according to IUPAC.

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