

Synthesis, Antimicrobial and Molecular docking studies of novel Benzimidazole derivatives

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Abstract: 2-(2-Chlorophenyl)-1H-benzo[d]imidazole derivatives were prepared by condensing 2-(2-Chlorophenyl)-1H-benzo[d]imidazole with different heterocycles. The synthesized derivatives were characterized by IR, ¹H NMR, ¹³C NMR, mass and elemental analysis. The compounds were screened for in vitro and microbial activity against panel of selected gram positive and gram negative bacterial strains using Ciprofloxacin as standard and molecular docking done with a Biotin carboxylase from E. Coli PDB CODE(3JZI).

Keywords: 2-(2-Chlorophenyl)-1H-benzo[d]imidazole; Antimicrobial agents; ciprofloxacin; Molecular docking:

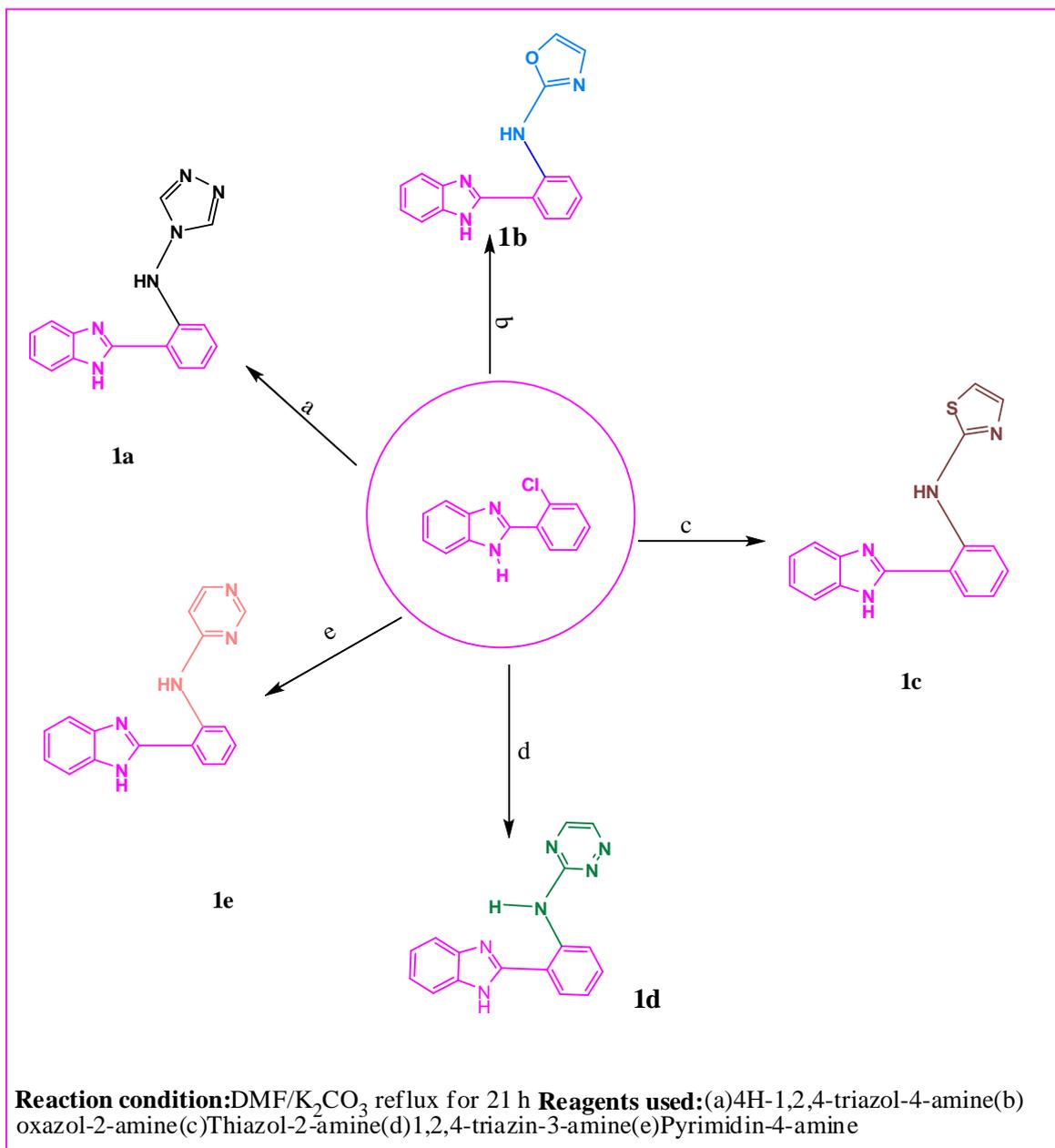
I. Introduction

Benzimidazole derivatives have found applications in diverse therapeutic areas including antitumor [1,2], antiulcer [3], anti-inflammatory [4], antiviral [5,6], antihelmintics [7,8], antibacterial [9,10] and antifungal [11,12] properties. Optimization of substituent around the benzimidazole nucleus has resulted in many drugs like albendazole, mebendazole, thiabendazole as antihelmintics; omeprazole, lansoprazole, pantoprazole as proton pump inhibitors and many lead compounds in a wide range of other therapeutic areas [13]. The widespread interest in benzimidazole containing structures has prompted extensive studies for their synthesis. Recently the interest in benzimidazole chemistry has been revived by the discovery that the 5,6-dimethyl benzimidazole moiety is part of the chemical structure of vitamin B12 [14].

Though all seven position in the benzimidazole nucleus can be substituted with variety of chemical entities, but most of the biologically active benzimidazole based compounds bear functional groups at 1,2 and/or 5 (or 6). Based on the above observations, we have planned to synthesize a novel series of benzimidazole derivatives derived from: 2-(2-Chlorophenyl)-1H-benzo[d]imidazole followed by their In-Vitro antibacterial activities. As an inception, various 2-(2-Chlorophenyl)-1H-benzo[d]imidazole derivatives were synthesized and characterized by FT-IR, NMR (¹H, ¹³C), mass, CHN analysis and the antimicrobial activities were screened. We used newly synthesized active inhibitors to expose the binding cavity of Biotin carboxylase from E. Coli by using Schrodinger suite program.

II. Results and Discussion

The synthesis of various benzimidazole derivatives were carried out as depicted in scheme-1. A broad band at 3413-3435 cm⁻¹ is ascribed to N-H stretching frequency of the imidazole moiety. A strong band at 3057 cm⁻¹ is due to the Aromatic (C-H) stretching frequencies. Hence the IR data illustrate the formation of the 2-(2-Chlorophenyl)-1H-benzo[d]imidazole derivatives. A sharp singlet at 12-13 ppm is assignable to -NH proton of benzimidazole and also peak at 4.7 ppm show the presence of CH₂. On focusing the ¹³C NMR spectral assignments, the signals at 164 ppm is due to C-N of heterocyclic compounds (1b-1e).



scheme-1

2.1. Antimicrobial studies

The antimicrobial activities of the synthesized compounds against different pathogens were determined by Agar Well diffusion method. Using sterile inoculation loop, 20 pure colonies of the test organism are transferred to 5ml of sterile nutrient broth and incubated at 37°C overnight for 18hrs. The modified agar well diffusion method of Perez et al. [15] was employed. Each selective medium was inoculated with the microorganism suspended in sterile water. Once the agar was solidified, it was punched with a six millimeters diameter wells and filled with 50µg/ml of the sample and blanks (ethanol and antibiotic). The test was carried out by triplicate. The plates were incubated at 35 ± 2°C for 24 h. The antimicrobial activity was calculated by applying the expression in µg/ml. The antibacterial activities in terms of minimum inhibitory concentration (MIC) of compounds (1a-1e) are depicted in Table-1.

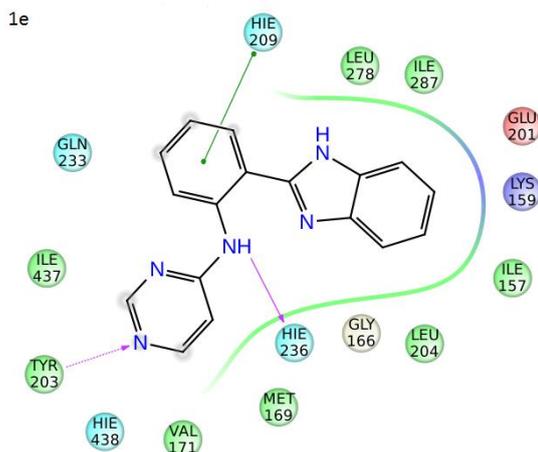
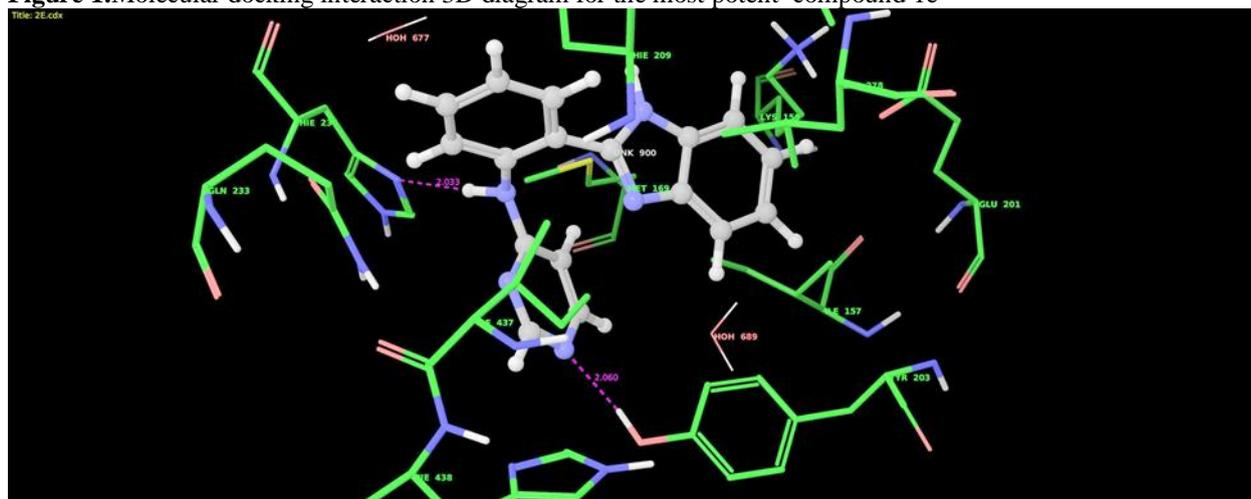


Figure 1. Molecular docking interaction 3D diagram for the most potent compound 1e



III. Experimental Section

Melting point (mps) were determined by open capillary method and are uncorrected. IR spectra were recorded by Jasco FTS 3000 HX(KBr pellets). ¹H NMR spectra were recorded on Bruker ADVANCE III NMR spectrometer (500 MHz) using TMS as internal standard (Chemical shifts in ppm). ¹³C NMR spectra were recorded on the same instrument at 125.76 MHz and are referenced using the central line of the solvent signal (DMSO-d₆ septet at δ = 39.5 ppm). Mass spectra were recorded with JOEL ac MATE II instrument. Elemental analysis (C, H and N) were performed with a Perkin Elmer 2400 series II CHN Analyzer. **General procedure for synthesis of compounds (1a-1e)[14]:**

N-(2-(1H-benzo[d]imidazol-2-yl)phenyl)-4H-1,2,4-triazol-4-amine(1a); 2-(2-Chlorophenyl)-1H-benzo[d]imidazole (2.2867g, 0.01mol) and K₂CO₃ (0.02mol, 2.76g) were stirred at room temperature in dimethyl formamide (25 ml) for half an hour and pinch of KI was added. After that 4H-1,2,4-triazole-4-amine (0.840g, 0.01mol) was added to reaction mixture which was refluxed for 21 hours until TLC showed completion of reaction. The reaction mixture was poured into water (20ml) and the mixture was extracted with ethylacetate. The organic extracts were washed with water, dried over anhydrous sodium sulphate and concentrated to obtain crude product. The residue was crystallised from diethyl ether to give pure compound. at room temperature. Physical data of compounds (1a-1e) are presented in **Table-3**. Begie yellow solid, Yield (35%); mp 350°C (dec); IR (KBr) 3339(NH Str), 3057 (-CH Str), 1613(aromatic, C=C str), 1184(C-N) cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ: 12.272(s, 1H), 8.219 (s, 2H), 7.56-7.23(m, 6H), 7.09 (s, 1H), 6.94-6.79(m, 2H), ¹³C NMR (125.76 MHz, DMSO-d₆) δ: 154.29, 146.9, 144.2, 128.6, 125.2, 123.1, 119.8, 116.5, 115.8, 115.6, 107.64; Mass(m/z): 276.11

N-(2-(1H-benzo[d]imidazol-2-yl)phenyl)oxazol-2-amine(1b); 2-(2-Chlorophenyl)-1H-benzo[d]imidazole (2.2867g, 0.01mol) and K₂CO₃ (0.02mol, 2.76g) were stirred at room temperature in dimethyl formamide (25 ml) for half an hour and pinch of KI was added. After that oxazol-2-amine (0.840g, 0.01mol) was added to reaction mixture which was refluxed for 36 hours until TLC showed completion of reaction. The reaction mixture was poured into water (20ml) and the mixture was extracted with ethylacetate. The organic extracts were washed with water, dried over anhydrous sodium sulphate and concentrated to obtain crude

product. The residue was crystallized from diethyl ether to give pure compound. at room temperature. yellow solid, Yield (45%); mp 432⁰C (dec); IR (KBr) 3358 (N-H Str), 3013 (-CH Str), 1608 (aromatic,C=C str) ,1180 (C-N) cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ: 12.23(s,1H),10.6 (s,1H),7.6(d,1H),7.56-7.23(m,5H),7.169(d,1H),7.160(m,1H)6.94-6.76 (m,2H),¹³ CNMR (125.76 MHz, DMSO-d₆) δ:164.8,154.2,146.9,144.2,128.6,125.2,123.0,119.8, 116.5, 115.8,115.6,107.64;Mass(m/z):276.29

N-(2-(1H-benzo[d]imidazol-2-yl)phenyl)thiazol-2-amine(1c):2-(2-Chlorophenyl)-1H-benzo[d]imidazole (2.2867g, 0.01mol) and K₂CO₃ (0.02mol,2.76g) were stirred at room temperature in dimethyl formamide (25 ml) for half an hour and pinch of KI was added. After that thiazole-2-amine (1.001g, 0.01mol) was added to reaction mixture which was refluxed for 35 hours until TLC showed completion of reaction. The reaction mixture was poured into water (20ml) and the mixture was extracted with ethyl acetate. The organic extracts was washed with water, dried over anhydrous sodium sulphate and concentrated to obtain crude product. The residue was crystallized from diethyl ether to give pure compound. at room temperature. yellow solid, Yield (50%); mp498⁰C dec); IR (KBr) 3334 (N-H Str), 3050(-CHStr),1608(aromatic,C=Cstr),1186(C-N)cm⁻¹; ¹H NMR(500MHz,DMSO-d₆)δ: 12.23(s,1H),10.5(s,1H),

,7.7(d,1H),7.56-7.09(m,6H),)6.94-6.76 (m,2H),6.72(dm1H);¹³CNMR(125.76MHz,DMSO-d₆)δ: 160.1,153.4,,148.9,142.4,137.8,129.8,121,119,115..5,115.3,113.6,108;Mass(m/z):292.36.

N-(2-(1H-benzo[d]imidazol-2-yl)phenyl)-1,2,4-triazin-3-amine(1d):2-(2-Chlorophenyl)-1H-benzo[d]imidazole (2.2867g, 0.01mol) and K₂CO₃ (0.02mol,2.76g) were stirred at room temperature in dimethyl formamide (25 ml) for half an hour and pinch of KI was added. After that 1,2,4-triazin-3 – amine (0.960g,0.01mol) was added to reaction mixture which was refluxed for 45 hours until TLC showed completion of reaction. The reaction mixture was poured into water (20ml) and the mixture was extracted with ethyl acetate. The organic extracts was washed with water, dried over anhydrous sodium sulphate and concentrated to obtain crude product. The residue was crystallized from diethyl ether to give pure compound. at room temperature. yellow solid, Yield (68%); mp 555⁰C (dec); IR (KBr) 3413 (-NH Str), 3041 (-CH Str), 1609(aromatic ,C=C str) cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ: 12.8 (s, 1H) ,9.7 (s,1H),8.2 (d,2H),7.56-7.09 (m,6H),6.94-6.79 (m,2H);¹³ CNMR (125.76 MHz, DMSO-d₆)δ:160.41, 153.4,149.89,146.6,141.45,138.53, 129.8,125.4,122.5,121.0,119.0,115.5,115.3,113.6,108.0;Mass(m/z):288.11

N-(2-(1H-benzo[d]imidazol-2-yl)phenyl)pyrimidin-4-amine(1e):2-(2-Chlorophenyl)-1H-benzo[d]imidazole (2.2867g, 0.01mol) and K₂CO₃ (0.02mol,2.76g) were stirred at room temperature in dimethyl formamide (25 ml) for half an hour and pinch of KI was added. After that pyrimidin-4-amine(0.950g,0.01mol) was added to reaction mixture which was refluxed for 45 hours until TLC showed completion of reaction. The reaction mixture was poured into water (20ml) and the mixture was extracted with ethyl acetate. The organic extracts was washed with water, dried over anhydrous sodium sulphate and concentrated to obtain crude product. The residue was crystallized from diethyl ether to give pure compound. at room temperature. yellow solid, Yield (68%); mp 495⁰C (dec); IR (KBr) 3323 (-NH Str), 3049 (-CH Str), 1608(aromatic ,C=C str) cm⁻¹; ¹H NMR (500 MHz, DMSO-d₆) δ: 12.35 (s, 1H)9.54 (s,1H), 8.82 (s,1H), 8.50(d,2H), 8.26(d,2H), 7.56-7.04(m,6H) 6.94-6.76(m,2H);¹³ CNMR (125.76 MHz, DMSO-d₆)δ:168.7, 159.6, 158.6,154.3, 148.9,146.6, 142.4,137.8,129.8, 125.4,122.5,121.0, 119.0, 115.5,113.6,108.0, 107.1;Mass (m/z): 287.12

IV. Conclusions

Fivenew 2-Chloromethyl-1H-benzimidazole derivatives weresynthesized in reasonably good yields. They were characterized by IR, ¹H,¹³C NMR(1D,2DNMR), mass and elemental analysis. All the newly synthesized compounds were tested for antimicrobial activity by agar well diffusion method. Among the screened samplescompound 1a exhibited as most active against E.Coli compared to other synthesized compounds.

Conflicts of Interest

“The authors declare no conflict of interest”.

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