

Synthesis, Characterization and Pharmacological evaluation of some Cinnoline (Furan) derivatives

Mishra Pankaj, Saxena Vikas, Kesheri Minu, Saxena Abhishek
*Department of Pharmaceutical chemistry Shridhar university pilani Rajasthan India

Abstract: The review of literature showed that cinnoline derivatives were found to elicit many pharmacological actions like anti-hypertensive, antithrombotic, antihistamine, antileukemic, CNS activity, anti tumor, antibacterial and antisecretory activity.

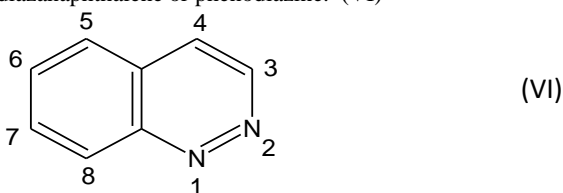
Furan can be found in many other drugs such as dacarbazine, metronidazole, cimetidine, flumazenil, thyroliberin, methimazole, pilocarpine and etomidate which are used as antineoplastic antibiotic, antiulcerative, benzodiazepine antagonist, prohormone, antihyperthyroid, muscarinic receptor.

In the substituted Cinnoline furan series, the compounds which are halogen mainly Chloro Substituted were showed potent antibacterial, anti-inflammatory and anti-fungal activity than other compounds. However Methyl substituted compound also Showed more potent antimicrobial activity and anti-inflammatory activity

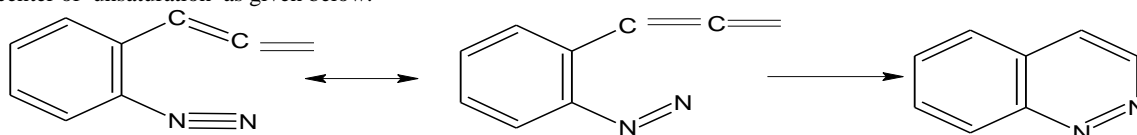
Keywords: cinnoline, furan, anti-inflammatory activity.

I. Introduction

Cinnolines: Cinnoline is a pale yellow solid, m.p. 24-25°C and was first discovered by Von Richter in 1883. He also prepared a cinnoline derivative from 2-aminophenylpropionic acid via intramolecular cyclization of the diazonium salt. The review of literature showed that cinnoline derivatives were found to elicit many pharmacological actions like anti-hypertensive, antithrombotic, antihistamine, antileukemic, CNS activity, anti tumor, antibacterial and antisecretory activity. They are reactive by virtue of the presence of a benzene ring and the electrophilic attack takes place in this ring. Cinnolines are the six-membered heterocyclic compounds having two hetero atoms in the ring. They are also called as 1, 2-benzodiazine or benzopyridazine or 1, 2-diazanaphthalene or phenodiazine. (VI)



The main approach for the synthesis of cinnoline is electrophilic attack by diazonium cation on carbon – carbon center of unsaturation as given below.



Furan:

Furan is a [heterocyclic organic compound](#), consisting of a five-membered [aromatic](#) ring with four carbon atoms and one oxygen. The class of compounds containing such rings are also referred to as furans. The parent compound is typically derived by the [thermal decomposition](#) of [pentose](#)-containing materials, cellulosic solids especially pine-wood. Furan is a colorless, [flammable](#), highly [volatile](#) liquid with a [boiling point](#) close to room temperature. It is [toxic](#) and may be [carcinogenic](#). Catalytic hydrogenation (see [redox](#)) of furan with a [palladium catalyst](#) gives [tetrahydrofuran](#). The name furan comes from the Latin *furfur*, which means [bran](#). The first furan derivative to be described was [2-furoic acid](#), by [Carl Wilhelm Scheele](#) in 1780. Another important derivative, [furfural](#), was reported by [Johann Wolfgang Döbereiner](#) in 1831 and characterised nine years later by [John Stenhouse](#). Furan itself was first prepared by [Heinrich Limpricht](#) in 1870, although he called it *tetraphenol*.

Objectives

1. Synthesis of new series of substituted cinnoline derivatives condensed with Imidazole Moieties.
2. Characterization of newly synthesized compounds by analytical and spectral methods viz., IR spectra, NMR spectra and Mass spectra.
3. Anti-inflammatory activity of some of the synthesized compounds.

II. Review Of Literature

Narayana et al., (2006) Studied antibacterial and antifungal studies on some new acetylcinnolines and cinnolinyl thiazole derivatives. They are reported antibacterial and antifungal activity.

Iradyan MA et al. (2007) reported potent anti-tumor activity in some amino imidazole compounds.

Saxena et al., (2009) Synthesis, characterization and biological activities of substituted cinnoline culphonamides. They are reported that these derivatives particularly halogen substituted cinnoline derivatives showed potent antimicrobial activity.

Dutta et al., (2010) Synthesized a series of 2-substituted-4,5-diphenyl imidazoles by refluxing benzil with different substituted aldehydes and screened for anthelmintic activity. The compounds showed significant anthelmintic activity compared to the standard drugs.

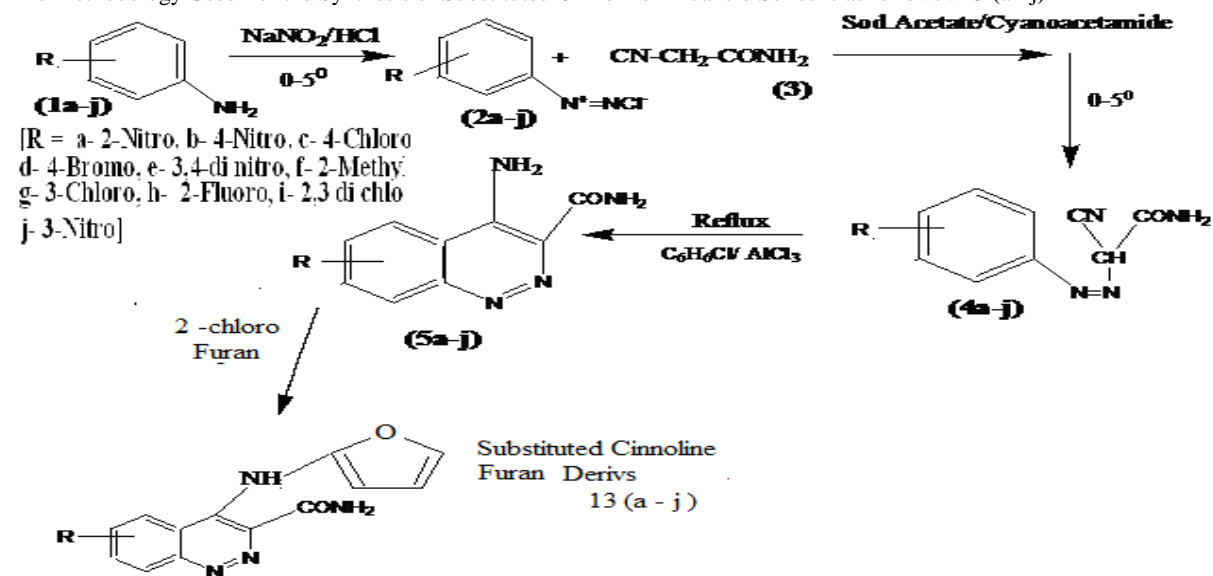
Patel et al., (2011) Studied synthesis and microbial evaluation of pyrazoline derivatives. They are reported that anti microbial activities.

Priyadarsini et al., (2012) prepared new substituted pyrazoles from o-hydroxyacetophenone and cinnamic acids as starting material through 1,3-diketones as intermediates. These intermediates on reaction with hydrazines in alkaline media produce pyrazoles. The antimicrobial activity of synthesized pyrazoles. In the most cases having Chloro substitution on the styryl ring was found to be more efficient.

Chaudhary et al., (2014) Studied synthesis and biological screening of some cinnoline derivatives. They are reported that newly synthesized compounds were screened for their anti-inflammatory and antibacterial activity.

III. Material And Method

The Methodology Used for the Synthesis of Substituted Cinnoline imidazole Series is as follows : 13 (a - j)



The synthesis of substituted cinnoline Imidazole derivatives by the described above method remitted in products with good yield.

Table : 3.0. Physical data of substituted 4(-2-amino-furan) cinnoline-3-carboxamide derivatives: (13a -j)

Compound Name	Com. No	Physical nature	M.P(°C)	Yield (%)
8-Nitro-4(-2-amino-2- furan) cinnoline-3-carboxamide	13DSD _a	Light Brick crystals	182-183°C	67.39%
6-nitro- 4(-2-amino- furan)cinnoline-3-carboxamide	13DSD _b	Yellow crystals	178-179°C	83.87%
6-chloro-4(-2-amino- furan) cinnoline-3-carboxamide	13DSD _c	Pale green Crystals	212-214°C	63.17%
6-bromo-4(-2-amino- furan) cinnoline-3-carboxamide	13DSD _d	Parrot Color crystals	202-204°C	72.92%
6,7 di nitro- 4(-2-amino- furan) cinnoline-3-carboxamide	13DSD _e	Dark Brownish red	196-198°C	81.05%
8-methyl- 4(-2-amino-2- furan) cinnoline-3-carboxamide	13DSD _f	Light red crystals	206-208°C	56.58%
7 chloro- 4(-2-amino- furan) cinnoline-3-carboxamide	13DSD _g	Violet green crystals	234-236°C	50.44%
8-Fluoro- 4(-2-amino- furan) cinnoline-3-carboxamide	13DSD _h	Orange yellow crys	205-207°C	45.23%
7,8-Di-chloro-4(-2-amino- furan) cinnoline-3-carboxamide	13DSD _i	Light Yellow cryst	261-263°C	49.77%
7- Nitro- 4(-2-amino-furan) cinnoline-3-carboxamide	13DSD _j	Light Brown Crystals	188-190°C	75.38%

Methodology For Anti-inflammatory:

The anti-inflammatory activity was assessed by rat paw edema method wherein the procedure of plethysmographic measurement of edema produced by planter injection of 1% w/v formalin in the hind paw of the rat was followed. The method described by Wilhelm and Domez as modified by Sisodia and Rao was used for measuring the paw volume. Suspension of phenylbutazone containing 40 mg/ml of drug was prepared in 2% gum acacia and used as standard drug. Suspensions of test compounds at a concentration of 40 mg/ml were also prepared in 2% gum acacia. The dose concentration of both standard drug and the test compounds was 100 mg/kg body weight. 1% w/v of formalin solution prepared and 0.1 ml of it in each case was injected in the planter region of left hind paw of albino rats.

Albino rats of either sex weighing 150-200 grams were used and divided into groups of six albino rats in each group. First group served as control, second group was used for standard drug phenylbutazone and the remaining groups served for compounds under investigation. An identification mark was made on both the hind paws just beyond tibiotarsal junction so that every time the paw was dipped in mercury column upto a fixed mark to ensure constant paw volume. Immediately after 30 minutes of drug administration, 0.1 ml of 1% w/v formalin was injected in the planter region of left paw of the rats. The right paw was used as reference for non inflamed paw for comparison. The paw volume of all the test animals was measured after 2nd and 4th hours of drug administration. The percentage of increase in edema over the initial reading was also calculated. The increase in edema of animals treated with standard test compounds were compared with the increase in the edema of untreated control animal with the corresponding intervals of 2nd and 4th hours. Thus the percentage inhibition of edema at known intervals in treated animals was calculated as given below .

$$\text{Percentage inhibition} = \frac{V_c - V_t}{V_c} \times 100$$

V_c = volume of paw edema in control animals

V_t = volume of paw edema in treated animals

Data analysis

The data were subjected to analysis of variance (ANOVA) as per statistical methods using SPSS (1996) software package. All the Synthesized compounds have shown anti-inflammatory activity to a certain extent as compared to standard drug Phenylbutazone. Among the tested compounds **15DSDc, 15DSDf and 15DSDi** have shown good activity by formalin induced rat paw edema method.

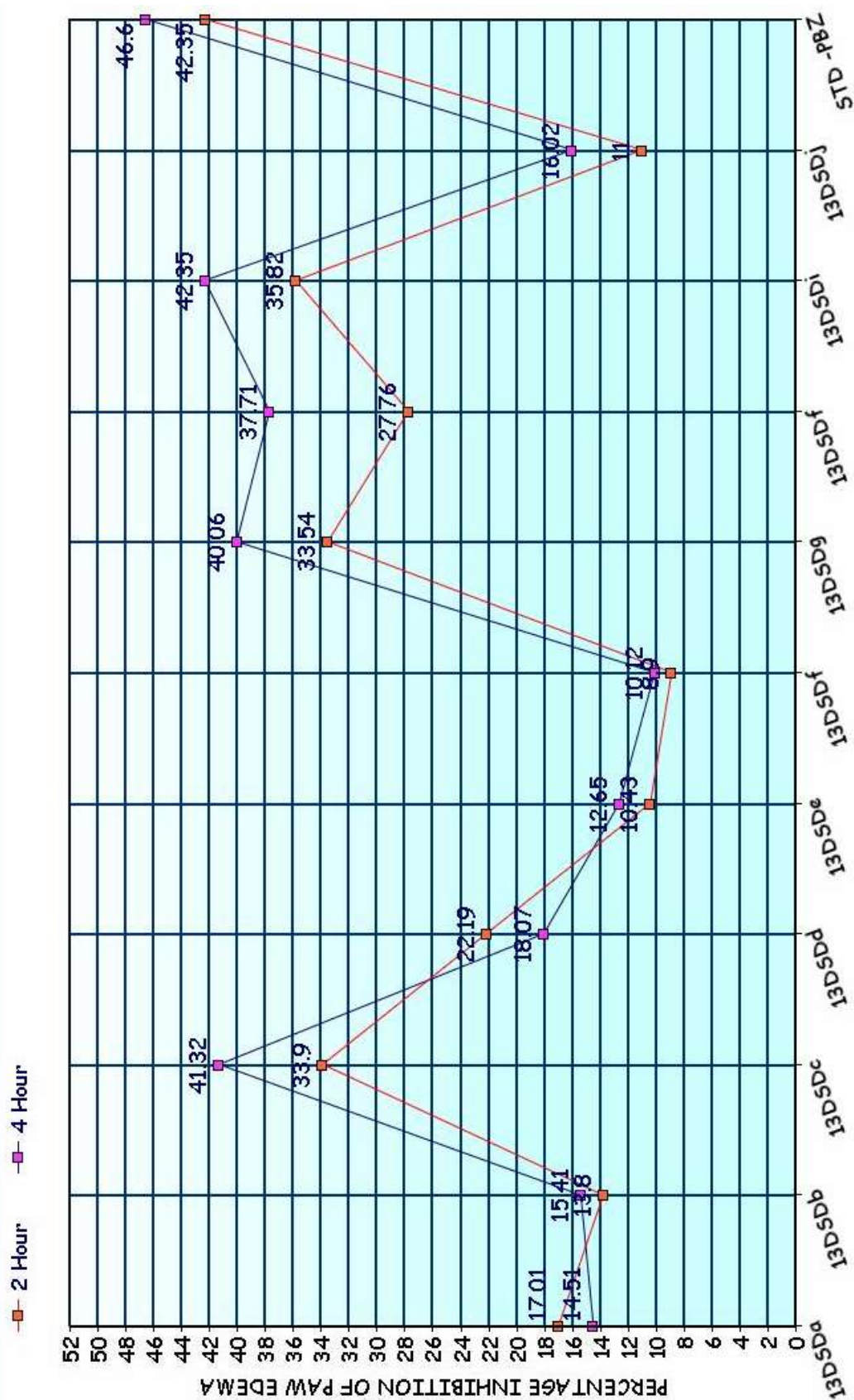
Result of Anti-inflammatory Activity

Compound	Substitution	Dose Mg/kg	Mean value (+S.E) of edema at different intervals		Percentage inhibition At Different intervals	
			2nd Hour	4th hour	2nd hr	4th hr
13DSDa	8 -Nitro	100	1.39 (±0.032)	1.30 (±0.001)	17.01	14.51
13DSDb	6- Nitro	100	1.45 (±0.019)	1.39 (±0.003)	13.80	15.41
13DSDc	6- Chloro	100	1.12 (±0.004)	1.02 (±0.004)	33.90	41.32
13DSDd	6-Bromo	100	1.35 (±0.039)	1.42 (±0.003)	22.19	18.07
13DSDe	6,7- di nitro	100	1.60 (±0.032)	1.53 (±0.021)	10.43	12.65
13DSDf	8- Methyl	100	1.64 (±0.0004)	1.59 (±0.006)	8.90	10.12
13DSDg	7 -Chloro	100	1.12 (±0.005)	1.03 (±0.004)	33.54	40.06
13DSDh	8-Fluoro	100	1.30 (±0.001)	1.07 (±0.012)	27.76	37.71
13DSDi	7,8- DiChloro	100	1.10 (±0.007)	1.01 (±0.032)	35.82	42.35
13DSDj	7- Nitro	100	1.51 (±0.002)	1.38 (±0.001)	11.00	16.02
Phenyl butazone	Standard	100	1.01 (±.003)	0.88 (±0.043)	42.35	46.6

IV. Result and Discussion

In the substituted Cinnoline furan series, the compounds which are halogen mainly Chloro Substituted were showed potent antibacterial, anti-inflammatory and anti-fungal activity than other compounds. However Methyl substituted compound also Showed more potent antimicrobial activity and anti-inflammatory activity.

COMPARATIVE PERCENTAGE INHIBITION OF INFLAMMATION



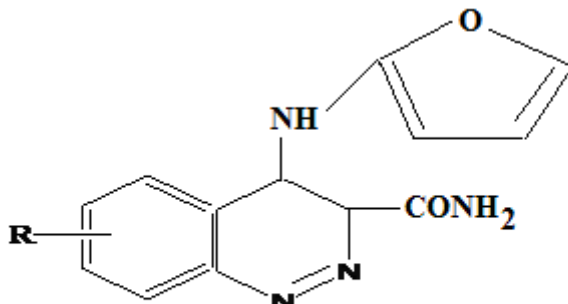
SUBSTITUTED CINNOLINE FURAN DERIVATIVES

In the present work the representative products were characterized by their infrared (IR) spectra, proton magnetic resonance (PMR) spectra and mass spectra. Some intermediates were characterized by measuring their melting point and comparing with literature value, wherever possible. The IR spectra were recorded by NICOLETT-IMPACT-400FT-IR

SPECTRO PHOTOMETER using a thin film supported on KBr pellets. The PMR spectra were recorded on JEOL-JMS D-300 (300 MHz) NMR spectro meter. All spectra were obtained in Deuterated Methanol and chemical shift values are reported as values in ppm relative to TMS ($\delta = 0$) as internal standard. Mass spectra were recorded on JEOL SX102 MS System operating at 70 ev.

Sample 15DSDi :

C.No. 15DSD_i _ 7,8-Di-chloro-4-(5-amino- Imidazole) cinnoline-3-carboxamide



**SUBSTITUTED CINNOLINE FURAN
DERIVATIVES**

➤ **IR (KBr) in cm^{-1} (Figure-13.1)**

Peak at 3466.1 cm^{-1} corresponds to NH stretching

Peak at 3341.5 cm^{-1} corresponds to asymmetric NH_2 group.

Peak at 3236.2 cm^{-1} corresponds to CH stretching.

Peak at 1631.9 cm^{-1} corresponds to C = O stretching.

Peak at 1470.6 cm^{-1} corresponds to aromatic C = C stretching.

Peak at 1113.0 cm^{-1} corresponds to C = S stretching.

Peak at 861.6 to 1298.5 corresponds to furan

➤ **$^1\text{H-NMR } \delta$ in ppm (figure.13.2)**

δ 7.96– 8.01 (2H, d, of cinnolines)

δ 7.4– 7.52 (3H,m, Furan)

δ 11.20 (1H, s, of NH)

δ 9.9 – 10.01 (2H, s, of CONH_2)

➤ **Mass in m/z (Figure-13.3)**

Molecular ion peak at $m/z = 326$ mHz is because of molecular formula $\text{C}_{13}\text{H}_{10}\text{Cl}_2\text{N}_4\text{O}_2$. Base peak is at $m/z = 160$ mHz. Fragment ion peak is observed at $m/z = 312$ because of $\text{C}_{13}\text{H}_{10}\text{Cl}_2\text{N}_3\text{O}_2$, $m/z = 241$ because of $\text{C}_9\text{H}_4\text{Cl}_2\text{N}_3\text{O}$, $m/z = 83$ because of $\text{C}_4\text{H}_4\text{ON}$.

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