

## Review on niosomal structure Through nasal Route

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**Abstract:** Nanotechnology has created one of the most dynamic science and technology domains at the confluence of physical sciences, molecular engineering, biology, biotechnology and medicine. There has been a considerable research interest in the area of developing drug delivery using nanoparticles (NP's) as carriers for small and large molecules. Targeting delivery of drugs to the diseased lesions is one of the most important aspects of drug delivery system especially brain. They have been used in-vivo to protect the drug entity in the systemic circulation, restrict access of the drug to the chosen sites and to deliver the drug at a controlled and sustained rate to this site of action. Niosomes are non-ionic surfactant vesicles obtained on hydration of synthetic nonionic surfactants, with or without incorporation of cholesterol or other lipids. They are vesicular systems similar to liposomes that can be used as carriers of amphiphilic and lipophilic drugs. Various polymers have been used in the formulation of niosomes for drug delivery research to increase therapeutic benefit, while minimizing the side-effects. It is obvious that niosome appears to be a well preferred drug delivery system over liposome as niosome being stable and economic. Also niosomes have great drug delivery potential for targeted delivery of anti-cancer, anti-infective agents. Drug delivery potential of niosome can enhance by using novel concepts like proniosomes, disomes and aspasome. Niosomes represent a promising drug delivery module. Niosomes are thoughts to be better candidates drug delivery as compared to liposomes due to various factors like cost, stability etc. Various types of drug deliveries can be possible using niosomes like targeting, ophthalmic, topical and parenteral.

**Key Words:** Nanotechnology, niosomes, nasal route, targeting.

Date of Submission: 20-01-2018

Date of acceptance: 19-02-2018

### I. Introduction

The drug delivery by nasal route generated the interest of the widespread among the community of the scientists as it is a good alternative route of administration that can avoid the first pass effect of some drugs that are susceptible for the enzymatically degradable drugs. Intranasal route also allows better absorption of the drug through the vascularity and permeability of the nasal mucosa. (1) The barriers that face the drug through the intranasal route is the enzymes that located in the nasal mucosa lining. Despite that, large numbers of the drug that include peptides, protein, vaccines and hormones can be delivered to the patient through the nasal route. Intranasal route of administration has several advantages and disadvantages as shown in table (2).

**Table ( 1):** The advantages and limitations of intranasal route(1).

Advantages	Limitations
1. Avoid the enzymatic degradation in GIT.	1. The required volume to be delivered into the nasal route is 20-200 ml.
2. Avoid first pass effects.	2. The components that have high molecular weight cannot be delivered.
3. Rapid onset of the drug.	3. Enzymes barriers in the nasal mucosa.
4. Use lower doses because of higher bioavailability.	4. Nasal irritation.
5. The route is noninvasive.	5. Limited mechanisms understanding.
6. Easily self-medication.	6. Large variability.
7. Transport the required dose directly to the systemic circulation and to the brain	7. Side effects by pathological activities.
8. Avoid over-doses.	
9. Not complex.	

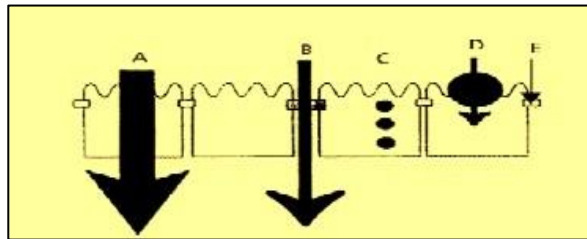
To overcome some limitation of the intranasal route we can formulate the drug into niosomes, pro-liposomes and films that enhance the drug permeability that is showed to be better than suspensions, solutions, sprays, emulsions, snuffs and ointments. The cause is that niosomes, pro-liposomes and films allow prolonged contact with the mucus membrane (1).

#### Nasal cavity functional feature:

The nasal mucosa is richly supplied with the blood to achieve the functions of the cavity of the nose that include humidification, heating, mucociliary clearance, olfaction and immunological functions. The surface area of the cavity is about 150160 cm<sup>2</sup>; that is the result of presence about 400 microvilli/cell the volume of these secretions of the nose about 15 ml/day.

All of these conditions favor the large permeability of the drug through the nose.(1)

**Drug permeation mechanism(2):**



**Figure (1)** : show different pathways as (A) show the drug pass through epithelium,

(B) paracellular transport, (C) transcellular transport, (D) carrier mediated transport and (E) intracellular tight junction

The permeation through the nasal cavity while administration of the drug can be carried out by either paracellular pathway as showed in figure 1 passively or actively and passively via transcellular pathway. In which it depends mainly on the lipophilicity of the drug.(1)

**II. Factors That Affect The Drug Permeability Through Nasal Route**

**Mucocilliary Clearance:**

It involves the combined actions of the mucus layer and the cilia and is an important factor in the physiological defence of the respiratory tract against inhaled hazardous particles. The composition, function and clinical aspects of nasal mucus have been widely reviewed. It is assumed that the speed of mucocilliary clearance in healthy humans is about 5 mm/ min; although this is easily influenced by pharmaceutical excipients, airborne irritants or diseases. The tips of the cilia are in contact with and transport the superficial viscoelastic mucus layer towards the nasopharynx, while the less viscous lower layer of the mucus is relatively stationary. Several workers, using various in vitro or in vivo methods, have investigated ciliary beat frequency in order to evaluate the effects of drugs or formulation additives or of infections in the upper airways on the mucocilliary system. The cilia beat in a coordinated fashion, with a frequency of approximately 10 Hz, when measured in in vitro studies on human nasal cilia (2).

**Enzymes:**

While nasal administration of drugs does avoid first pass hepatic metabolism, there is a broad range of metabolic enzymes situated in the nasal mucosa which can limit the bioavailability of some drugs, especially those containing peptides or proteins. Among the enzymes present are the oxidative phase I enzymes (e.g. cytochrome P-450 enzymes), non-oxidative enzymes, conjugative phase II enzymes and proteolytic enzymes such as endo- and exo-peptidases. The nasal enzyme population and/or activities vary extensively among different species. However, the level of activity seems to be lower for nasal enzymes than for those in the gastrointestinal tract or liver, on the basis of the amount of tissue involved.(3)

**Nasal pathophysiology various pathophysiological changes:**

Such as the common cold, seasonal rhinitis, nasal polyps and cancer, may also alter absorption from the nasal cavity in different ways, although this has not yet been thoroughly investigated; While it has been demonstrated that a rhinovirus infection in vitro causes sloughing of epithelial cells and destruction of the epithelial layer, microscopy studies of mucosal biopsies from otherwise healthy patients with colds didn't show any abnormalities in ciliated cells.(2)

**Physico-chemical characteristics of the substance:**

The physicochemical characteristics of the administered drug, which can influence nasal absorption, include molecular weight, solubility, dissolution rate, charge, partition coefficient, pKa, particle size and the presence of polymorphism.(4)

**Solubility of the drug in nasal secretion:**

As the drug needs to be solubilized in the nasal secretion before its permeation. Diurnal variation as diurnal rhythm affects the nasal secretions. As the secretion and clearance rates are reduced at night which affects the drug permeation.

**Ph Of The Nasal Cavity:**

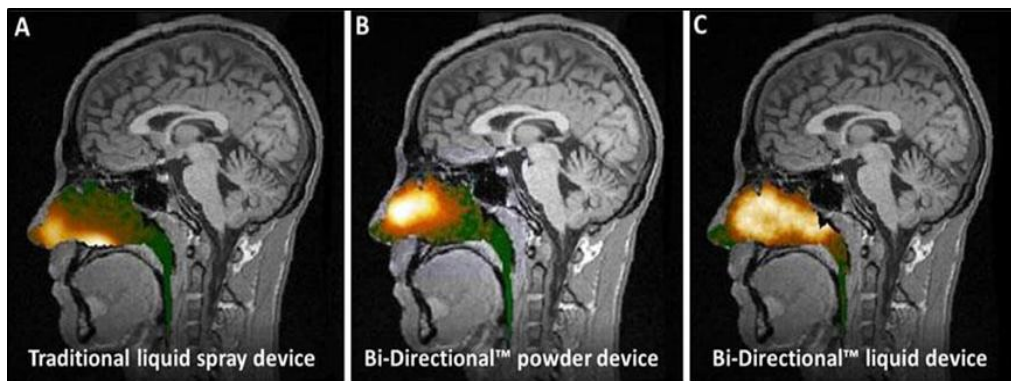
As it is varied from 5.5 to 6.5 in adults and from 5 to 7 in infants. Usually drug permeation increased when the nasal pH is lower than the drug pKa because under this condition the drug molecules present as unionized species. Change in nasal pH affects the ionization of the drug molecules that affect the drug permeation.(1)

**Molecular weight:**

An inverse relationship between molecular weight and percent absorption has been reported; data are supported by the results of rat studies compiled with literature data, which indicate good bioavailability for compounds with molecular weights up to 1000 kDa in formulations without adjuvants). (10) However, contrary to the findings of Donovan et al. (1990) no difference in absorption characteristics between gastrointestinal and nasal mucosae was found in rats (6) Accordingly, mechanisms other than the suggested aqueous pores between cells of the nasal mucosa) might be involved in the absorption of large molecules. Other studies have demonstrated that hydrophobicity is an important factor in nasal drug delivery in contrast to studies on quaternary ammonium compounds where a decrease in absorption was found with increased lipophilicity and molecular weight .(3)

**The drug administered through nasal route taken by different methods:**

**Figure (2):** show the drug deposition two minutes after drug delivery using (A) traditional liquid spray, (B) breathpowered Bi-directional powder device and (C) breathpowered Bi-directional liquid device incorporating spray pump



**Devices for liquid formulations:**

Liquid formulations include emulsion, aqueous solutions and suspensions. Liquid formulation considered to be good method as it does humidification in which counteract the dryness of the nose that usually accompany the chronic diseases of nose. The devices used with liquid formulations mainly spray pump systems as showed in figure (2) usually require preservatives which is mainly benzalkonium chloride that may cause reduction in ciliary movement and irritation also recent studies proved that the use of benzalkonium chloride in long term use is well tolerated and safe for the long-term use.(11)

**Drops delivered with pipette:**

It is the oldest method for the nasal delivery in which the breast milk has been dripped for the treatment of nasal congestion in infants, methanol vapors have been used to wake fainted people and both vapors and drops are still present in the market till now. Drops are administrated by liquid sucking into glass dropper; the dropper is inserted into the nostril with extended neck then squeezing the rubber top to allow the drops to be emitted. For the purposes of multiple use drops have been replaced by metered dose spray pump.(11)

**Liquid delivery by rhinyle catheter and squirt tube:**

It is a simple way in which physicians used to insert the drug into the nose by inserting a tip of catheter to the required area under visual control then squirt the required drug to the desirable location. This method mainly used in the studies that include the use of animals. But this method has a dangerous disadvantage that it cannot be used for self-medication.(12)

**N Squeeze bottles:**

This method used to deliver OTC drugs as topical decongestant. As the air filled plastic bottle is partly squeezed

in which the drug is ionized while delivering from a jet outlet. The size and particles of the dose varied according to the applied force. While releasing the pressure the nasal secretions and micro-organisms are sucked into the bottle so the use of the squeeze bottles is not recommended for children.(11)

**Spray pumps metered dose:**

Which offers high reproducibility of the dose that emitted and plume geometry in tests applied in vitro. The size of the particles and the plume geometry varied within limits and depend on the pump properties, the orifice of the actuator, formulation and the force applied. In the traditional spray pump the air is replaced by emitted liquid so preservatives are required to prevent contamination. Pump manufacturers are developing other methods to avoid the use of preservatives.(11)

**Single and due dose spray devices:**

It is preferred to be used for narrow index drugs and used also for vaccines and other drugs that need single accurate dose.(11,13)

**Nasal pressurized metered dose inhalers:**

Most of the drugs used for the local action delivered using spray pumps but some other drugs as nasal aerosols taken by inhalers (11,12)

**Devices for powder formulations:**

Powder sprayers with compressible compartment to supply pressure in which when it is released it make a plume of powder particles which is like that produced by liquid spray.(11) Breath actuated inhalers in which the human use his own breath for the powder inhalation into the nostril from a capsule or blister.(11) Nasal insufflator describes the devices that contain nose piece and mouth piece which are connected. The drug delivery carried out while exhalation of the drug into the mouth piece to close the velum, then the airflows carries the particles of powder into nose through the nose piece of the device.(11)

**Nasal physiology that affect the drug delivery**

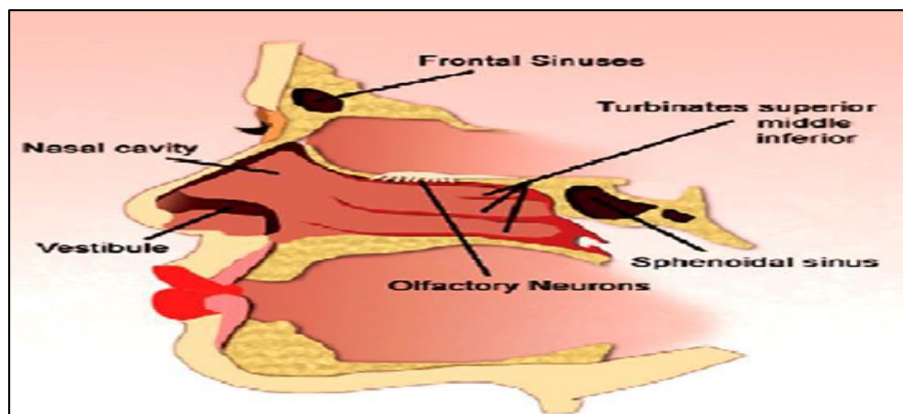


Figure ( 3): Nasal anatomy

**Aerodynamic and nasal valve:**

The dynamic segment of narrow anterior triangular of the nasal anatomy called nasal valve which is the limiting segment primary flow and extends anterior and posterior to the inferior turbinate head approximately 23cm from the opening of nostril. The narrow triangular slit plays a role of dynamic valve for the modification of the rate and the direction of the airflow during the respiration.(12)

**The nasal mucosal clearance and filtration:**

The anterior region to the valve is called vestibule which is lined by nonciliated squamous epithelium then the valve is gradually converted into ciliated epithelium which is typical of ciliated respiratory epithelium posterior to the region of the valve. Behind the valve of the nose, the turbinate of the nose divided the nasal cavity into slit like passages with larger surface area and cross sectional area. The speed of the laminar airflow is slowed down to 23m/s and disrupted with eddies that promote deposition of the particles that are carried with air and behind the region of the valve. The ciliated respiratory mucosa which is posterior to the nasal valve, covered by a blanket designed protective mucosa that traps particles and micro-organisms.(11).

### **Thenasal cycle:**

The physiology of the alternation between congestion and decongestion observed has been observed in about 80% of the humans that are healthy is called the nasal cycle. The autonomic cycle is changed during air flow resistance which mainly depends on blood content of submucosal capacitance vessels which constitute the erectile component at sites which are critical, notably the region of nasal valve. The erectile tissue of septal and lateral walls and turbinates are responding to stimuli including sexual and physical activity and the state of the emotion that can override and cause modification to basic cyclic rhythm. Because of the cycle one nostril more congested to the other most of time and major air flow is passing from one nostril while the other one is remaining narrow. (14)

### **Nasal sinus vasculature and lymphatic system:**

The site of deposition affects the route and extent of administration in which the targeted organ distributes as shown in figure (3). The branches of the maxillary and ophthalmic arteries supply the mucus membrane that covers the sinus, meatuses, septum and turbinate. Where the superior labial branch of facial artery which supply the part of septum in the vestibular region. The turbinate is located in the lateral nose wall that is highly vascularized with blood flow and then acts as a radiator to the airway. (12) Substances that are absorbed from the region of the anterior seem to drain by jugular veins, where drugs absorbed from mucous behind the valve of nose seem to drain by veins that are reached to sinus cavernous as the venous blood become indirect contact with the carotid arteries walls. (12)

### **Targeted nasal delivery:**

In most cases the broad distribution of the active constituents of the drug on the surface of the mucosa appears good for the drug that required for the local action, Vaccination or systemic circulation. In case of nasal polyposis and chronic sinusitis the delivery targeted to superior and middle meatuses where the polyps originate appears desirable and sinus opening present. Another exception is the drug that are delivered from nose to the brain are more targeted to the upper part of the nose where the olfactory nerves are present. (20)

### **Brain targeting through intranasal route:**

When the drug passes the barriers of the nasal epithelium and enters the nose submucous space. The drug molecule diffused across the arachnoid membrane then enters the CSF compartment of olfactory region and then distributed with the flow of CSF and absorbed into the circulation. The drug that are lipophilic with molecular weight <400 Da will pass the blood brain barriers by lipid mediation and also passing the nasal epithelium barrier and arachnoid membrane. And the other drugs that are water soluble (hydrophilic) or have molecular weight about >400 Da, then the drug pass the membranes poorly by free diffusion so in this case it is essential to disrupt the epithelial barriers to complete drug transportation. (16,25) The drug that do not pass the blood brain barriers by lipid medication that is approximately about % of the present drugs in the market, can pass the BBB by carrying it to be transported on the endogenous transporters. Small and large molecules of the drug must be formulated to allow the transportation on the blood brain barriers endogenous transporters. The BBB endogenous transporters divided into three categories carrier mediated transport, receptor mediated transport and active efflux transport. (16,25)

### **Nano-technology based drug delivery system:**

Nanotechnology is changing the perception of drug administration using conventional dosage forms. The term of nanoparticles refer to a particulate drug delivery system where particle size is in nanometer range (1-1000 nm). Nano particles are being investigated extensively in order to develop drug delivery systems capable of allowing penetration through physiological barriers. (24) Over the past decades, there has been considerable research interest in the area of developing nano technology by using nano particles as carriers for small and large molecules. Various polymers have been used in the formulation of nano particles. This review presents the most outstanding contributions in the field of nanotechnology. The word 'Nano' is derived from Latin word, which means dwarf. Nano size refers to one thousand millionth of a particular unit thus nanometer is one thousand millionth of a meter (i.e. 1n=10<sup>9</sup>m). The term nanotechnology has been most commonly used in the fields of science like electronic, physics and engineering since many decades. However, bio medical and pharmaceutical fields remain yet to be explored. (24)

### **Major advantages of nano sizing include:**

- [1] Increase surface
- [2] Enhanced solubility
- [3] Increase rate of dissolution and oral bio availability
- [4] Rapid onset of action
- [5] Less amount of dose required in the field of pharmacy.

For applications to medicine and physiology these materials and devices can be designed to interact with a high degree of functional specificity, thus allowing a degree of interaction between technology and biological systems not previously attainable. (18)

### III. Types of Nanotechnology-Based drug delivery systems

#### Polymeric Nanoparticles (Nps)

Nanoparticles are defined as particulate dispersions or solid particles with diameters ranging from 1 to 1,000 nm. By using several methods, nanoparticles have been prepared, these methods include ionic gelation or coacervation, polymerization and emulsion solvent evaporation (24), solvent diffusion or spontaneous emulsification, supercritical fluid technology, spray drying, nanoprecipitation, and particle replication in non-wetting templates (10). The mechanism by which transport of nanoparticles can pass the BBB can be explained by the improved adsorption of the nanoparticles to the capillary walls due to the increased retention of the nanoparticles in the brain blood capillaries. This resulted in a higher concentration gradient, which increases the transport across the endothelial cell layer and enhances the delivery to the brain. Using poly sorbate 80 as the coating agent caused inhibition of the efflux system which can also facilitate the transport of drug. Nanoparticles may lead to limited permeabilization of the brain endothelial cells due to induction of local toxic effects on the brain vasculature. Enhancement of permeability of drug across the BBB can be improved by solubilizing lipids of the endothelial cell membrane by the use of surfactant. Through the tight junctions, nanoparticles could permeate the BBB, which are open between the endothelial cells of the brain blood vessels. Another method to facilitate the drug delivery to the brain is through endocytosis by the endothelial cells followed by the drug release within these cells. In order to improve the delivery of drug to the brain, nanoparticles can be administered through the intranasal route. Other methods to improve the delivery include antibodies or polymers to improve nasal absorption, coating nanoparticles with polyethylene glycol. The retention time of nanoparticles delivered through the intranasal route can be increased by surface modification of the NPs with mucoadhesive polymers. (17,18)

#### Solid lipid carriers

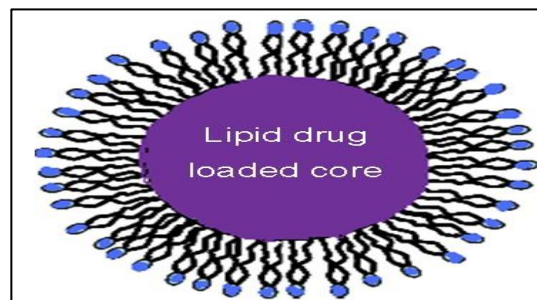


Figure (4): structure of solid-lipid nanoparticle (32)

Solid lipid nanoparticles (SLNs) as shown in figure 10 are typically spherical particles, with average diameters between 10 and 1,000 nm when dispersed in water. SLNs have a core matrix of solid lipid that can solubilize lipophilic molecules (Müller et al 2002). The lipid core, typically consisting of steroids (eg, cholesterol), monoglycerides (eg, glycerol mono stearate), diglycerides (eg, glyceryl behenate), triglycerides (eg, tristearin), waxes (eg, cetyl palmitate) or fatty acids (eg, stearic acid), is stabilized by surfactants through the combination of emulsifiers which help in preventing agglomeration of particles (30). Solid lipid nanoparticles are prepared from an emulsifier, lipids and water or solvent by using different methods which include: the solvent evaporation method, ultrasonication/high shear technique, high pressure homogenization, the supercritical fluid method, the spray drying method, the ME based method, the solvent emulsification and diffusion method, the precipitation technique or the double emulsion method. The use of SLNs or nanocarriers lipids for the drug delivery to the brain can facilitate the problem of crossing the BBB, as these formulations can be used intranasally to bypass the BBB or penetrate directly the BBB. Promoting electrostatic interactions with mucus in addition to mediating the adsorptive-mediated transcytosis of cationic NPs across the BBB can be done through the use of cationic lipids that improve mucoadhesion in the nasal cavity. The transport of nanoparticles coated with surfactant across the BBB takes place through endocytosis mediated by the endothelial cells of the brain capillaries (30).

#### Surfactant-based systems:

Drug delivery systems in which molecules of surfactant self-aggregate which takes place in the presence of water and lead to the formation of structures with different parameters that differ with respect to concentration. These drug delivery systems are called surfactant-based drug delivery systems. These self-aggregated surfactants become more organized even when other components such as oils or other surfactants are added to the system of surfactant and water. Thus, microemulsion (MEs) and nanoemulsions (NEs) can be produced. Microemulsions are usually thermodynamically stable isotropic liquids formed by water, mixing oil, and surfactants all together. In contrast, nanoemulsions are conventional emulsions that contain very small particles. The sizes of droplets of microemulsion range between 10 and 140 nm, which lead to the formation of systems that are optically transparent and thermodynamically stable. Nanoemulsions are not transparent with diameters ranging up to 140 nm, so they are less thermodynamically stable than microemulsion (29). The two systems are very different because ME phases are formed by self-assembly and NEs are formed by mechanical shearing. Other parameters can differ

guish MEs from NEs: MEs are more stable in long term storage than NEs; MEs can be agitated, cooled, or heated and then return to their original conditions, whereas NEs cannot return to their original conditions (29); Microemulsion resulted from spontaneous mixtures of water, oils and surfactants, however in order to facilitate the formation of microemulsion it is recommended to apply heating or stirring because of kinetic energy barriers that must be overcome. Nanoemulsions can be formed by use in part of some external energy provided by microfluidizers, high pressure homogenizers and sonication methods to convert the mixture into a colloidal dispersion or phase inversion. Nanoemulsion formed from curcumin undergone development for drug delivery through intranasal route, and the results from behavioral experiments showed improved learning and memory in the group treated with curcumin-loaded nanoemulsion compared with the group treated with the pure drug. (34)

### Vesicular drug delivery system:

Systems that have been presented as carriers for delivery of drug for many years are called Vesicular drug delivery systems. They were supposed to achieve many goals which included: enhancement of the transport of drug through different biological membranes, targeted drug delivery and controlling the release of drug as well as prolonging this release if needed. These systems include liposomes, transfersomes, niosomes, vesosomes, ethosomes, and colloidosomes (15). Only vesicles employed in the nasal drug delivery will be discussed such as

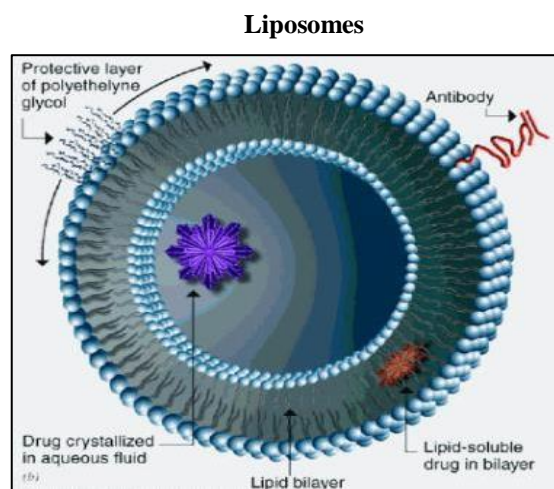


Figure (5): structure of liposome (17)

Liposomes as shown in figure 1 are microscopic vesicles spherical in shape composed of one or more bilayers of lipids, arranged around a central core aqueous in nature (21). They are made of nontoxic, biodegradable, and natural constituents such as phospholipids. Liposomes may include in their composition membrane stabilizers such as cholesterol as well as small amounts of charging agents and because of this desirable composition liposomes have the ability to encapsulate drugs with widely differences in the lipophilic feature of drugs where the lipophilic portion being located in the bilayer of lipid and the portion which is hydrophilic is located in the central aqueous core. Liposomes can change the pharmacodynamics and pharmacokinetics of the entrapped drugs and it is considered an efficient drug delivery system. The activity of liposomes as carriers for drug depends upon different factors such as release rates, stability, and efficiency of encapsulation, distribution in body after administration, rigidity and size surface charge. The properties of liposomes can be varied and controlled by changing the preparation methods and using different types of lipids. Poor stability of liposomes is considered the main problem in the research of liposomes and this problem resulted from the chemical degradation of components of liposomes as well as the problems of physical stability (28), that can include the size change upon storage and the loss of entrapped drug. Increasing the rigidity of the bilayer can decrease the loss of entrapped material membrane also reducing the water content of liposome formulations producing the so-called proliposomes can help in improving the physical stability of liposomes. (30)

### Proliposomes:

Proliposomes are free-flowing, dry particles that upon addition of water form vesicular dispersions instantly. Their free-flowing particulate properties allow the fabrication of these nanoaggregates into solid dosage forms, which then is converted to liposomes on contact with water or biological fluids. Proliposomes are prepared by penetrating a solution of drugs and phospholipids in volatile organic solvents into the microporous matrix of water-soluble carrier particles, then the evaporation of the organic solvent takes place. Phospholipids and drugs are thus deposited in the microporous structure of the carrier materials, thus keeping the free-flowing surface characteristics of the carrier materials. Proliposomes have some advantages which included decreasing the problems of physical instability such as leakage, fusion and aggregation. In addition, proliposomes are characterized by ease of storage, distribution, transportation and dosage. (19)

## Niosomes

Niosomes are multilamellar vesicular structure of non-ionic surfactants as shown in figure 12, similar to liposomes however instead of phospholipids that are a component of liposomes they are composed of non-ionic surfactants. (22)

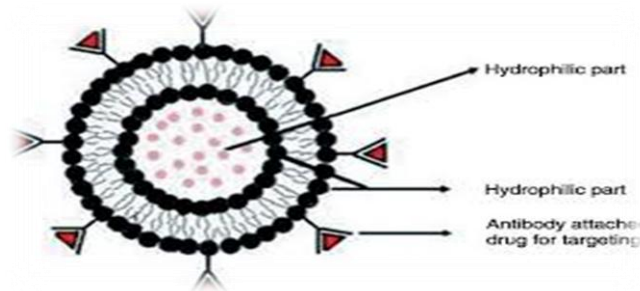


Figure (6): structure of niosomes (26)

## Advantages of Niosomes (11)

- Niosomes have the ability to entrap lipophilic, hydrophilic and amphiphilic drugs so they can be used to deliver wide range of drugs with different properties.
- Niosomes have better therapeutic effect when compared to other conventional oily formulations and better patient compliance.
- Composition, size, shape and fluidity of niosomes drug can be controlled.
- Niosomes due to depot formation can provide controlled and sustained release of drugs.
- Niosomes has the advantage of drug targeting to different organs. (6)
- Niosomes characterized by higher bioavailability than other conventional dosage forms.
- Niosomes can be administered through different routes such as topical, oral, parenteral and intranasal.
- Niosomes have the ability of increasing the permeation of drug through the skin.
- Niosomes protect the active constituent from acid and biological enzymes that lead to increase the stability of the active constituent.
- Niosomes are stable as they are composed of non-ionic surfactants (more than liposomes that composed of phospholipids).
- Transportation, handling and storage of niosomes is easy.
- Niosomes can improve Oral bioavailability of the drug.
- Niosomes are non-immunogenic (safe) to the body, biodegradable and biocompatible. (10)

## IV. Factors affecting the physiological properties of niosomes

### Membrane Additives:

The number of additives added to formulation of niosomes besides the non-ionic surfactant and the drug can increase the stability of niosomes, also permeability and morphology of vesicles are influenced by numbers of additives such as: increasing the rigidity and stability of niosomes by adding cholesterol to the formulation. (10)

### Temperature of Hydration:

Hydration temperature has a great influence on size and shape of niosomes. Composition of vesicles of niosomes is affected by the change in temperature of the system of niosome. The change in temperature can also cause transformation in vesicle shape. At temperature of 25°C Polyhydral vesicles of C<sub>16</sub>G<sub>2</sub>:solulan C<sub>24</sub>(91:9) is formed, however at temperature of 45°C it is converted into vesicles spherical in shape. (39)

### Properties of Drugs:

The entrapment efficiency of drugs in niosomes is affected by lipophilicity, chemical structure, hydrophilicity, molecular weight and the hydrophilic/lipophilic balance (HLB) of the drug. Charge on polymer may be increased as it is affected by interaction between drug particles and the head group of the non-ionic surfactant which in turn causes repulsion of the surfactant bilayer that finally leads to increase in the size of vesicle. (26)



#### **Amount and Type of Surfactant:**

The mean size of niosomes can increase with the increase of HLB value of surfactants like span 20 (HLB 8.6) to span 85 (HLB 1.8) in direct proportional relation, this phenomenon is due to decrease of surface free energy that associates the increase in HLB value thus increase in hydrophilicity of surfactant. The structure of bilayer is more disordered in the liquid state, while in the gel state alkyl chain is present in well-ordered structure. The phase transition temperature of the gel-liquid is used to characterize the lipids and surfactant, also the efficiency of drug entrapment is affected by phase transition temperature, for example span 60 provides better entrapment efficiency due to high value of HLB value of surfactant has a great effect on efficiency of entrapment of drug in niosomes for example at HLB value 8.6 niosomes have high entrapment efficiency but at HLB value 14 to 17 there is no suitability for formulation of niosomes (28).

#### **Cholesterol Content and Charge on the Surfactant:**

The presence of cholesterol content in the niosomal bilayer can increase the entrapment efficiency and the diameter of niosome. Cholesterol can act by two ways either by decreasing the chain order of the gel state bilayer or increasing the chain order of liquidated bilayer. It has been found that rigidity of bilayer increases and the drug release rate decreases because of the high concentration of cholesterol. (28)

#### **Method of Preparation of niosomes**

Method of preparations can also affect properties of niosomes for example Reverse phase evaporation as well as microfluidization method can be used to produce vesicles that are smaller in size and with greater stability. (32)

#### **Hand Shaking Method**

Initially surfactant and cholesterol are dissolved in some organic solvent such as benzene, chloroform and ether then the solvent is evaporated under decreased pressure in a rotary evaporator in a round bottom flask that leaves the mixture of cholesterol and solid surfactant on the walls of round bottom flask then using aqueous solution containing drug with continuous shaking, this layer was then rehydrated which causes swelling of the surfactant layer. Swelled amphiphiles eventually fold and form vesicles which entrap the drugs. (26).

#### **Ether Injection Method:**

A slow injection of a solution containing a certain ratio of surfactant and cholesterol in ether into aqueous solution of the drug that is preheated and maintained at 60 °C through the specified gauze needle. Formation of unilamellar vesicles of the surfactants containing the drug takes place due to vaporization of ether. Ether cannot be used in case of thermolabile drugs so in this case fluorinated hydrocarbons have been used as an alternative, because their vaporization takes place at a much lower temperature. The size of niosomes that result from this method ranges from 50 to 1000 nm, which mainly depend on the experimental conditions and formulation variables. (36)

#### **Sonication Method:**

In this method initially in the aqueous phase the cholesterol-surfactant mixture is dispersed, then at temperature at 60 °C this dispersion is probe sonicated for 10 minutes that leads to the formation of multilamellar vesicles (MLV). Which are further ultrasonicated either by bath sonicator or probe sonicator, which finally resulted in the formation of unilamellar vesicles. (26).

#### **Reverse Phase Evaporation Method:**

The solution of cholesterol and surfactant is prepared in a mixture of chloroform and ether (1:1). Then the aqueous solution of drug is added and at temperature four and five °C sonication occurred. Phosphate buffer saline (PBS) is added to the solution obtained from sonication and the solution is subjected to further sonication which leads to the formation of gel. Thereafter pressure is reduced and temperature is raised to 40 °C for the solvent removal. For ten min, the PBS is added again and heated on water bath at 60 °C to produce niosomes. (39)

#### **Microfluidization Method:**

Two fluidized streams (one containing the surfactant and the other one containing drug) interact at ultra high speed, in certain micro channels within the chamber of interaction in such a way that the energy supplied to the system remains in the area of niosomes formations. This is called submerged jet principle. It leads to reproducibility in the formulation of niosomes, smaller size of niosomes and better uniformity (29).

#### **Extrusion Method:**

A mixture of diacetyl phosphate and cholesterol is prepared by the use of rotary vacuum evaporator the solvent is evaporated leaving a thin film. The thin film is then hydrated with aqueous drug solution and the

suspension thus obtained is extruded through the polycarbonate membrane which has mean pore size of 0.1mm and then placed in series up to eight passages that lead to formation of uniform size niosomes.(38 )

## **V. Components Of Niosomes**

### **Non-ionic Surfactants:**

In bilayer lattice, non-ionic surfactants place themselves where the hydrophilic heads face the aqueous region while the lipophilic head orient themselves in such a way that the interaction with the aqueous media would be minimized (face the lipid bilayer). Every bilayer folds over itself as continuous membrane in order to achieve thermodynamic stability. Types of non-ionic surfactant include: Alkyl Esters such as Sorbitan esters which are the most preferred Surfactant used for the preparation of niosomes in this category, Alkyl Ethers such as monoalkyl glycerol ether, Fatty Acid and Amino Acid Compounds and Alkyl Amides such as glucosides have also been used to produce niosomal vesicles.(15)

### **Cholesterol:**

Cholesterol is mainly used for the formulation of niosomes and it is a steroid derivative that is essential component of cell membrane and affects the permeability as well as fluidity of bilayer (10). Properties of niosomes such as encapsulation efficiency, rigidity, membrane permeability, ease of rehydration of freeze dried niosomes and their toxicity are affected by incorporation of cholesterol. Cholesterol prevents the vesicle aggregation by electrostatic forces or repulsive steric that leads to the conversion from the gel to the liquid phase in the system of niosome; hence the niosome becomes less leaky in nature.

### **Charged Molecule:**

In order to increase the stability of niosomes, some charged molecules are added to niosomes which prevents aggregation of niosomes by electrostatic repulsion. The negatively charged molecules that are used include phosphotidic acid and diacetyl phosphate (DCP). Similarly, stearyl pyridinium chloride and stearylamine (STR) are the well-known positively charged molecules used in niosomal preparations.(1,16) Only 2.5—5 mol percentage concentration of charged molecules is tolerable because high concentration can inhibit the niosome formation.(39)

## **VI. Types of Niosomes**

### **Proniosomes:**

This type of niosomes containing surfactant and carrier that requires to be hydrated before its usage, this hydration leads to the formation of dispersion of aqueous niosome. Proniosomes decreases the aggregation, leaking and fusion problem associated with niosomal formulation.

### **Bola Surfactant Containing Niosomes:**

The surfactants that are made of omega-hexadecyl-bis-(1-aza-18 crown-6) (bola surfactant): span-80/cholesterol in 2: 3: 1 molar ratio.

### **Niosomes in Carbopol Gel:**

In this type, niosomes prepared from cholesterol, drug and spans are then incorporated in carbopol-934 gel (1% w/w) base containing glycerol (30% w/w) and propylene glycol (10% w/w).

### **Aspasomes:**

This type formed by Combination of highly charged lipid diacetyl phosphate, cholesterol and acorbyl palmitate.

### **Vesicles in Water and Oil System (v/w/o):**

This type of niosomes can be created by. Adding of niosomes suspension that is formed from mixture of solulan C24, cholesterol and sorbitol monostearate to oil phase at temperature 60 °C that leads to the formation of vesicle in water in oil (v/w/o) emulsion which can be converted to vesicle in water in oil gel (v/w/o gel) by cooling to room temperature, This gel has the ability to provide controlled release as well as entrap /proteinous drugs and proteins also protect it from enzymatic degradation after oral administration.

### **Niosomes of Hydroxyl Propyl Methyl Cellulose**

It the type of niosomes in which, a base including 10% glycerin of hydroxy propyl methyl cellulose was initially developed and then niosomes were incorporated in it.

**Table (2):** variation between niosomes and liposomes (Manda et al, 2014) on between niosomes and liposomes (28).

Liposomes	Niosomes
Requires special method for storage, handling and purification of phospholipids	No special methods are required for such formulations comparatively
Phospholipids are prone to oxidation	non-ionic surfactants are stable (Mahaleet al, 2012)
More expensive	Less expensive

## VII. Conclusion

Niosome are non-ionic surfactant vesicles obtained on hydration of synthetic nonionic surfactants, with or without incorporation of cholesterol or their lipids. They are vesicular systems similar to liposomes that can be used as carriers of amphiphilic and lipophilic drugs. Niosome are promising vehicle for drug delivery and being non-ionic; and Niosomes are biodegradable, biocompatible nonimmunogenic and exhibit flexibility in their structural characterization. Niosomes have been widely evaluated for controlled release and targeted delivery for the treatment of cancer, viral infections and other microbial diseases. Niosomes can entrap both hydrophilic and lipophilic drugs and can prolong the circulation of the entrapped drug in body. Encapsulation of drug in vesicular system can be predicted to prolong the existence of drug in the systemic circulation and enhance penetration into target tissue, perhaps reduce toxicity if selective uptake can be achieved. This review article focuses on the advantages, Disadvantages, preparation methods, factors affecting, characterizations, invitro methods, drug release kinetics, and applications of niosome.

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Dr: Dalia Abd Elaty Mostafa "Review on niosomalstructure Through nasal Route." IOSR Journal of Pharmacy and Biological Sciences (IOSR-JPBS) 13.1 (2018): 29-40.